

# A New Diprenyl Coumarin and Alkaloids from the Bark of *Zanthoxylum dimorphophyllum* (Rutaceae)

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The alkaloids chelerythrine, norchelerythrine, oxyavicine, canthine-6-one, 4,5-dihydrocanthin-6-one, and  $\gamma$ -fagarine were isolated from *Zanthoxylum dimorphophyllum* bark, together with two coumarins, scoparone and dimoxylin. This latter is a novel compound whose structure was elucidated on the basis of its spectral data.

## Introduction

The genus *Zanthoxylum* L. (Rutaceae) includes some 250 species of trees and shrubs, with a worldwide, but predominantly tropical distribution (Engler, 1896; Engler, 1931). Morphologically, it is the only truly choricarpous genus in the Rutaceae, with fully free and stalked carpels (Gut, 1966). The very unspecialized flower morphology and vascular supply suggest a primitive position within the family (Moore, 1936; Das Graças *et al.*, 1988). Previous chemical studies on the genus have shown it to be a rich source of secondary metabolites (Waterman and Grundon, 1983). Surprisingly, none of the Vietnamese species have been so far studied from a chemical point of view. In a continuation of our studies on Rutaceous plants (Tillequin, 1997), we report here the isolation and structure determination of a novel coumarin, dimoxylin, together with the identification of a coumarin and six alkaloids, from the bark of *Zanthoxylum dimorphophyllum* Hemsl. (Hemsley, 1895) collected in Viêt Nam.

## Results and Discussion

Eight secondary metabolites were isolated from the CH<sub>2</sub>Cl<sub>2</sub> extract of *Zanthoxylum dimorphophyllum* bark. Three were identified as the benzophenanthridine alkaloids chelerythrine, norchelerythrine, and oxyavicine, widespread in the Rutaceae family (Krane *et al.*, 1984). Other

known alkaloids included canthine-6-one (Ohmoto *et al.*, 1976), 4,5-dihydrocanthin-6-one (Rosenkranz and Schmid, 1968), and the furo[2,3-*b*]quinoline  $\gamma$ -fagarine (Narasimhan and Mali, 1974). The two remaining compounds were scoparone (6,7-dimethoxycoumarin) (Joseph-Nathan *et al.*, 1984) and a novel coumarin, dimoxylin.

Dimoxylin (**1**) was obtained as a white amorphous product. The empirical formula was determined by accurate mass measurement as C<sub>20</sub>H<sub>28</sub>O<sub>6</sub>. The UV spectrum recorded in MeOH was suggestive of a 7-oxygenated coumarin. The IR spectrum showed characteristic bands at 3415 and 1712 cm<sup>-1</sup> accounting for alcoholic hydroxyl groups and for the pyrone-carbonyl, respectively. In the aromatic region, the <sup>1</sup>H NMR spectrum displayed a pair of doublets (*J* = 9.5 Hz) at 6.21 and 7.58 ppm typical for a coumarin unsubstituted in the pyrone ring, whereas a singlet at 7.28 ppm was consistent with the presence of three substituents in the aromatic ring. At higher field, typical signals accounted for one aromatic methoxyl and two different C-prenyl derived substituents, a 2,3-dihydroxy-3-methylbutyl side chain, and a 3-hydroxy-3-methylbutyl group. The assignment of the methoxyl and prenyl moieties was carried out using multi-impulsional HETCOR and COLOC experiments. Of particular interest were the following three bond COLOC connectivities: i) H-1'a at 2.57 ppm and H-1'b at 2.90 ppm to C-5 at 127.7 ppm ii) H-5 at 7.28 ppm to C-4 at 143.9 ppm,

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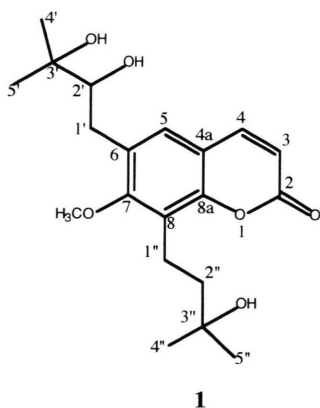
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C-7 at 159.8 ppm, C-8a at 151.7 ppm, and C-1' at 32.3 ppm. These data permitted to locate unambiguously the 2,3-dihydroxy-3-methylbutyl substituent at C-6, and hence the 3-hydroxy-3-methylbutyl chain at C-8. Therefore, the structure of dimoxylin can be depicted as **1**. The absolute configuration of the chiral center at C-2' could not be determined, due to the small amount of product isolated.



The isolation of coumarins together with that of alkaloids belonging to the benzophenanthridine, canthinone, and furoquinoline series indicates a lack of high skeletal specialization in the biosynthesis of *Zanthoxylum dimorphophyllum* secondary metabolites. From a chemotaxonomic point of view, it accounts for the primitive position of the genus *Zanthoxylum* in the Rutaceae family, in full agreement with a recent evolutionary interpretation of the family (Das Graças *et al.*, 1988).

## Experimental

### General experimental procedures

Optical rotation was obtained on a Perkin Elmer 241 polarimeter. Specific rotation ( $[\alpha]_D$ ) is reported in deg/dm. UV spectrum ( $\lambda_{\max}$  in nm) was recorded in spectroscopically grade MeOH on a PU 8700 Philips spectrophotometer. IR spectrum ( $\nu_{\max}$  in  $\text{cm}^{-1}$ ) was obtained from potassium bromide pellet on a Nicolet FT-IR 510 instrument.  $^1\text{H}$ -NMR ( $\delta$  [ppm],  $J$  [Hz]) and  $^{13}\text{C}$  NMR spectra were recorded at 300 MHz and 75 MHz, respectively, using a Bruker AC-300 spectrometer. Multi-impulsional 2D NMR experiments ( $^1\text{H}$ - $^1\text{H}$  COSY,  $^1\text{H}$ - $^1\text{H}$  NOESY,  $^{13}\text{C}$ - $^1\text{H}$  HETCOR, and  $^{13}\text{C}$ - $^1\text{H}$

COLOC) were performed using standard Bruker microprograms. High resolution mass spectrum (HR-MS) and Fast Atom Bombardment (FAB-MS) were recorded on a Micromass ZAB 2 – SEQ spectrometer. Mass spectra was recorded with a Nermag R 10–10C spectrometer, using desorption-chemical ionization (DCI-MS; reagent gas:  $\text{NH}_3$ ) technique. Column chromatographies were carried out with silica gel 20–45  $\mu\text{m}$ . Flash column chromatographies were conducted using silica gel 60 Merck (35–70  $\mu\text{m}$ ) with an overpressure of 300 mbars (Still *et al.*, 1978). Microanalyses were in agreement with calculated values  $\pm 0.4\%$ .

### Plant Material

Bark of *Zanthoxylum dimorphophyllum* Hemsl. was collected at Pà Cò, Mai Châu (Hoà Bình, Việt Nam), on 4 March 1997. A Voucher sample (VN 227) is kept in the herbarium of the Institute of Ecology of the National Center for Science and Technology in Hà Nội, Việt Nam.

### Extraction and Isolation

Dried, pulverized bark of *Zanthoxylum dimorphophyllum* (1 kg) was extracted with  $\text{CH}_2\text{Cl}_2$  (4  $\times$  2 l) at room temperature. The solvent was removed under reduced pressure to give a crude extract (15.3 g). An aliquot (2.5 g) was subjected to flash column chromatography on silica gel, using a  $\text{CH}_2\text{Cl}_2$ -MeOH gradient of increasing polarity to yield 12 fractions. Further column chromatographies on silica gel 20–45  $\mu\text{m}$ , performed on fractions 4 to 10, successively gave chelerythrine (34 mg), norchelerythrine (40 mg), scoparone (20 mg),  $\gamma$ -fagarine (28 mg), oxyavicine (22 mg), canthine-6-one (120 mg), dimoxylin (65 mg), and 4,5-dihydrocanthin-6-one (12 mg).

### Spectroscopic data

**Dimoxylin (1)**,  $[\alpha]_D^{20} +23.5$  (1 g/100 ml,  $\text{CH}_2\text{Cl}_2$ ); UV (MeOH)  $\lambda_{\max}$  (log  $\epsilon$ ) 217 (4.39), 252 (3.68), 294 (4.12), 325 (3.98) nm; IR (KBr)  $\nu_{\max}$  3415, 3055, 2972, 2936, 1712, 1606, 1567, 1470, 1396, 1134, 1051, 907, 828, 734  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 300 MHz)  $\delta$  1.27 (9H, s, 3 C- $\text{CH}_3$ ), 1.30 (3H, s, C- $\text{CH}_3$ ), 1.68 (2H, m,  $\text{CH}_2$ -2''), 2.57 (1H, dd,  $J$  = 14 Hz,  $J$  = 9 Hz, H-1'a), 2.76 (2H, m,  $\text{CH}_2$ -1''), 2.90 (1H, dd,  $J$  = 14 Hz,  $J$  = 1 Hz, H-1'b), 3.10 (3H, br.

s, D<sub>2</sub>O exch., 3 OH), 3.60 (1H, dd,  $J = 9$  Hz,  $J = 1$  Hz, H-2'), 3.78 (3H, s, O-CH<sub>3</sub>), 6.21 (1H, d,  $J = 9.5$  Hz, H-3), 7.28 (1H, s, H-5), 7.58 (1H, d,  $J = 9.5$  Hz, H-4); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz)  $\delta$  18.6 (C-1''), 23.7 (C-CH<sub>3</sub>), 26.0 (C-CH<sub>3</sub>), 28.8 (C-CH<sub>3</sub>), 29.0 (C-CH<sub>3</sub>), 32.3 (C-1'), 42.9 (C-2''), 61.8 (O-CH<sub>3</sub>), 70.5 (C-3''), 72.8 (C-3'), 78.2 (C-2'), 114.2 (C-3), 115.2 (C-4a), 124.1 (C-8), 127.7 (C-5), 130.0 (C-6), 143.9 (C-4), 151.7 (C-8a), 159.8 (C-7), 161.4 (C-2); HR-FAB-MS found: 387.1768 (calcd for [C<sub>20</sub>H<sub>28</sub>O<sub>6</sub> + Na]<sup>+</sup>, 387.1783); DCI-MS  $m/z$  382 [M+NH<sub>4</sub>]<sup>+</sup>, 365 [MH]<sup>+</sup>, 347.

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